

**Question 1:**

By studying a recently discovered species of bacteria, you find that the enzyme Zeta catalyzes the conversion of A to B. You purify Zeta and perform enzyme kinetics experiments, measuring  $V_{\max}$  to be  $60 \mu\text{M}/\text{min}$ ,  $K_M$  to be  $15 \mu\text{M}$ , and  $k_{\text{cat}}$  to be  $10^6/\text{sec}$ .

Unfortunately, you lose part of your notes before transcribing them to your notebook, so you **don't know** how much purified Zeta and [A] you added to the different tubes. Try to reconstruct the conditions of your original experiment, detailing your reasoning.

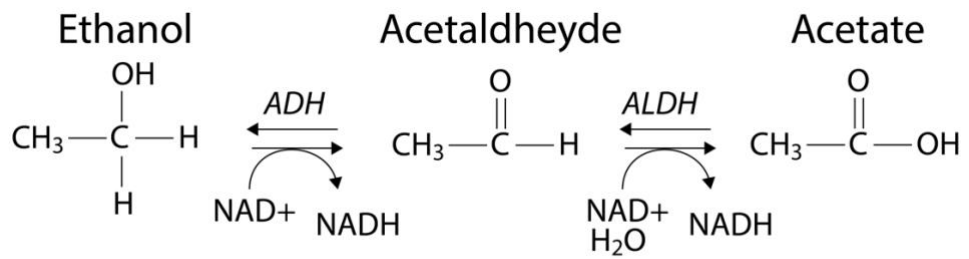
- (a) In one of your tubes, you measured the production of  $24 \mu\text{M}$  of B per minute. What was the concentration of [A] you added? [1 pt]
  - (b) What was the concentration of Zeta you added? [1 pt]
  - (c) You discover that your reaction contained  $1.5 \mu\text{M}$  of a known competitive inhibitor of Zeta with a  $K_i$  of  $3 \mu\text{M}$ . Now, knowing this, calculate the  $K_M$  for the reaction in the absence of the inhibitor. [2 pts]
- 

**Question 2:**

Alcohol dehydrogenases (ADHs) are enzymes that catalyze the oxidation of **primary** and **secondary alcohols** to **toxic aldehydes** and **ketones**, respectively. These are then converted to **less toxic acetate byproducts** by the enzyme **Acetaldehyde Dehydrogenase (ALDH)**.

- (a) In cases of methanol poisoning, **intravenous ethanol infusion** is used as treatment. Explain the principle behind this treatment. [1 pt]
- (b) The  $K_M$  of the liver ADH isozyme  $\alpha\beta 1$  for ethanol and methanol is  $1 \text{ mM}$  and  $15 \text{ mM}$ , respectively. Calculate the kinetic parameters for the **methanol-to-formaldehyde** reaction catalyzed by liver ADH isozyme  $\alpha\beta 1$  in the presence of  $1 \text{ mM}$  ethanol. [2 pts]
- (c) A researcher purifies a **new ADH isoform** and finds that it catalyzes the conversion of **0.30 g of ethanol** (**molecular weight = 46 g/mol**) per minute at **37 °C** at  $V_{\max}$ . What is the turnover number ( $k_{\text{cat}}$ ) of ADH (in  $\text{min}^{-1}$ )? Assume the **molecular weight of ADH** is **81,000 g/mol**. [2 pts]
- (d) **Alcohol flush reaction** is a condition in which a person develops **flushing** or **blotches** associated with **erythema** on the **face, neck, shoulders, ears**, and, in some cases, the **entire body** after consuming alcohol. Approximately **30–50%** of **East Asians** carry the **ALDH2\*2 allele** on **chromosome 12**, which is associated with **alcohol flush reaction**. **ALDH2\*2** bears a **lysine-for-glutamate** substitution at

position **487** within the **cofactor binding site**. By analyzing the reactions depicted in **Figure 3**, derive a **possible mechanism** by which **ALDH2\*2** contributes to alcohol flush reaction. **[2 pts]**



**Fig. 3**